

Characterization of vitamin B₁₂ compounds in biofertilizers containing purple photosynthetic bacteria

Tomohiro Bito¹, Noriharu Ohishi², Shigeo Takenaka³, Yukinori Yabuta^{1,2}, Emi Miyamoto⁴, Eiji Nishihara^{1,2} and Fumio Watanabe^{1,2,*}

¹Division of Applied Bioresources Chemistry, United Graduate School of Agricultural Sciences, Tottori University, Tottori, Japan, ²School of Agricultural, Biological and Environmental Sciences, Tottori University, Tottori, Japan, ³Department of Veterinary Science, Graduate School of Life and Environmental Sciences, Osaka Prefecture University, Osaka 598-8531, Japan, ⁴Department of Health and Nutrition, Nagasaki International University, Sasebo, Japan

ABSTRACT

There is currently no data available on the vitamin B₁₂ content of biofertilizers containing purple photosynthetic bacteria or whether these biofertilizers contain “true B₁₂” or inactive corrinoid compounds (or both). Therefore, we tested the vitamin B₁₂ content of three commercially available biofertilizers containing purple photosynthetic bacteria. No or traces of vitamin B₁₂ were found in two of the biofertilizers (A and B) tested, while the third (C) contained a considerable amount (53.5 µg /L). To evaluate whether biofertilizer C actually contained vitamin B₁₂ or other corrinoids inactive in humans, the extracted compounds were purified using an immunoaffinity column and then identified as vitamin B₁₂ (main) and/or inactive corrinoids (minor, Factor III) using liquid chromatography electrospray ionization mass spectrometry (LC/ESI-MS/MS).

KEYWORDS: biofertilizer, cobalamin, lettuce, purple photosynthetic bacteria, vitamin B₁₂

INTRODUCTION

Vitamin B₁₂ (B₁₂) or cobalamin is synthesized only by certain bacteria [1]. B₁₂ synthesized by

bacteria is concentrated mainly in the bodies of higher predatory organisms in the natural food chain. Foods derived from animals (i.e., meat, milk, egg, fish, and shellfish), but not plants are considered to be the major dietary sources of B₁₂ [2]. Thus, strict vegetarians have a greater risk of developing a B₁₂ deficiency compared to non-vegetarians [3]. The major symptoms of B₁₂ deficiency are neuropathy and megaloblastic anemia [4]. Therefore, there is a need to identify plant foods that contain high levels of B₁₂ to prevent vegetarians from developing a B₁₂ deficiency.

Mozafar [5] demonstrated that the addition of an organic fertilizer, cow manure, significantly increased the B₁₂ content (17.8 ng/g dry weight) in spinach leaves. However, our unpublished works indicate that most organic fertilizers, especially those made with animal manure, contain considerable amounts of inactive corrinoid compounds, which have also been reported to be present in human feces and account for more than 98% of total corrinoid compounds [6].

Biofertilizers are products containing living cells of beneficial microorganisms, which can accelerate and improve plant growth by providing nutritionally important elements (e.g., nitrogen and phosphorus) [7]. Although cyanobacteria are responsible for biological nitrogen (N₂) fixation in

*Corresponding author
watanabe@muses.tottori-u.ac.jp

flooded rice fields [8], they contain large amounts of pseudovitamin B₁₂ (pseudo-B₁₂) that is inactive in humans [9]. N₂-fixing purple photosynthetic bacteria were reported to have beneficial effects on plant growth [10] and the ability to synthesize B₁₂ *de novo* [11]. If biofertilizers containing purple photosynthetic bacteria contain a substantial amount of “true B₁₂,” these fertilizers would be useful for enriching B₁₂ in plants. However, there is no information on the B₁₂ content of biofertilizers containing purple photosynthetic bacteria or whether these biofertilizers contain “true B₁₂” or inactive corrinoid compounds (or both).

In this study, we determined the B₁₂ content of three biofertilizers containing purple photosynthetic bacteria and characterized the B₁₂ compounds found in these biofertilizers using liquid chromatography electrospray ionization mass spectrometry (LC/ESI-MS/MS).

MATERIALS AND METHODS

Materials

B₁₂ was obtained from Sigma (St. Louis, MO, USA) and 5-hydroxybenzimidazolyl cyanocobamide (Factor III) was provided by Dr. Stüpperich, Ulm University, Germany. The B₁₂ assay medium for *Lactobacillus delbrueckii* subspecies *lactis* (formerly *L. leichmannii*) ATCC7830 was obtained from Nissui (Tokyo, Japan). Silica gel 60 thin-layer chromatography (TLC) aluminum sheets were obtained from Merck (Darmstadt, Germany). A Shimadzu (Kyoto, Japan) UV-visible spectrophotometer (UV-2550) was used to measure the turbidity of *L. delbrueckii* test cultures with the microbiological B₁₂ assay method. Biofertilizers containing purple photosynthetic bacteria were purchased at local markets in Japan. All other reagents used were of the highest purity commercially available.

Extraction and assay of vitamin B₁₂

Each biofertilizer (1 L) was added to the same volume of 5 mmol/L acetate buffer (pH 4.5) containing 0.01% (w/v) potassium cyanide (KCN). Total B₁₂ was extracted by boiling this in a draught chamber (Dalton Co., Tokyo, Japan) for 30 min. B₁₂ was assayed at least five times independently by the microbiological method with

L. delbrueckii ATCC 7830. Since *L. delbrueckii* ATCC 7830 can utilize both deoxyribosides and deoxyribonucleotides (known as an alkali-resistant factor) as well as B₁₂ [12], the amount of B₁₂ was calculated by subtracting the values of the alkali-resistant factor from the total B₁₂ values.

Bioautography of vitamin B₁₂ compounds with vitamin B₁₂-dependent *Escherichia coli* 215

Bioautography of the B₁₂ compounds was done according to the method of Tanioka *et al.* [13]. After an aliquot of each B₁₂ extract of the biofertilizers was concentrated and partially purified with a Sep-Pak Plus C18 cartridge (Waters Corp.), concentrated (100 times) extracts, authentic B₁₂, and pseudo-B₁₂ (10 µg/L each) were spotted on a silica gel 60 TLC sheet and developed with 2-propanol/NH₄OH (28%)/water (7:1:2 v/v) in the dark at 25°C. Once the TLC sheet was dried, agar containing basal medium and precultured *E. coli* 215 was overlaid and then incubated at 30°C for 20 h. After being sprayed with a methanol solution of 2,3,5-triphenyltetrazolium salt on the gel plate, B₁₂ compounds were stained as red in color indicating *E. coli* growth.

LC/ESI-MS/MS analysis

The selected biofertilizer extracts were put on Sep-Pak[®] Vac 20cc (5 g) C18 Cartridges (Waters Corp.), which had been washed with 75% (v/v) ethanol and then equilibrated with distilled water. After the C18 Cartridge was washed with 30 mL of distilled water, B₁₂ compounds were eluted with 30 mL of 75% (v/v) ethanol. The eluate was evaporated to dryness under reduced pressure. The residual fraction was dissolved with 5 mL of distilled water and centrifuged at 10000 g for 10 min to remove insoluble materials. The supernatant fraction was loaded onto an immunoaffinity column [EASI-EXTRACT[®] B₁₂ Immunoaffinity Column (P80) R-Biopharm AG, Darmstadt, Germany] and then B₁₂ compounds were purified according to the manufacturer's recommended protocol.

The purified samples, authentic B₁₂, and Factor III were dissolved in 0.1% (v/v) acetic acid and filtered with a Nanosep MF centrifuge device (0.4 µm, Pall Corp., Tokyo, JAPAN) to remove small particles. An aliquot (2 µL) of the filtrate was then analyzed

using LCMS-IT-TOF coupled with an Ultra-Fast LC system (Shimadzu, Kyoto, JAPAN). The purified sample was injected in an InertSustain column (3 μ m, 2.0 x 100 mm, GL Science, Tokyo, JAPAN) and equilibrated with 85% solvent A [0.1% (v/v) acetic acid] and 15% solvent B (100% methanol) at 40°C. B₁₂ compounds were eluted with a linear gradient of methanol (15% solvent B for 0-5 min, 15%-90% solvent B for 5-11 min, and 90%-15% solvent B for 11-15 min) at a flow rate of 0.2 mL/min. ESI conditions were determined by injecting authentic Factor III or B₁₂ into the MS detector to achieve optimum parameters to detect parent and daughter ions of the B₁₂ compound. ESI-MS was operated in positive ion mode with argon as the collision gas. The identification of Factor III (m/z 672.272) and B₁₂ (m/z 678.292) representing [M+2H]²⁺ was confirmed by comparison of the observed molecular ions and the retention times.

RESULTS AND DISCUSSION

The amount of B₁₂ in three commercially available biofertilizers containing purple photosynthetic bacteria was determined using a microbiological method. No or traces of B₁₂ were found in two of the biofertilizers (A and B) tested while the third (C) contained a considerable amount (53.5 μ g /L) of the vitamin.

The B₁₂ compounds found in biofertilizer C were analyzed with the *E. coli* 215 bioautogram after being separated with silica gel 60 TLC (Fig. 1). A concentrated extract of biofertilizer C was identified as a clear single spot, with a R_f value identical to that of authentic B₁₂ but not to that of pseudo-B₁₂, which is inactive in humans. For biofertilizers A and B, no or an indistinct spot was detected. B₁₂ compounds were purified from the extracts of biofertilizer C with an immunoaffinity column and then identified by LC/ESI-MS/MS (Fig. 2). As described previously [14], authentic B₁₂ (C₆₃H₈₈CoN₁₄P; monoisotopic mass 1354.5674) was eluted as a peak with a retention time of 7.30 min. MS results of authentic B₁₂ indicated a major divalent ion of m/z 678.2937 [M+2H]²⁺, and isotope distribution data supported the determination that B₁₂ predominantly formed a divalent ion under the LC/ESI-MS conditions. MS/MS spectrum of B₁₂ indicated that the ion of m/z 359.0981 was

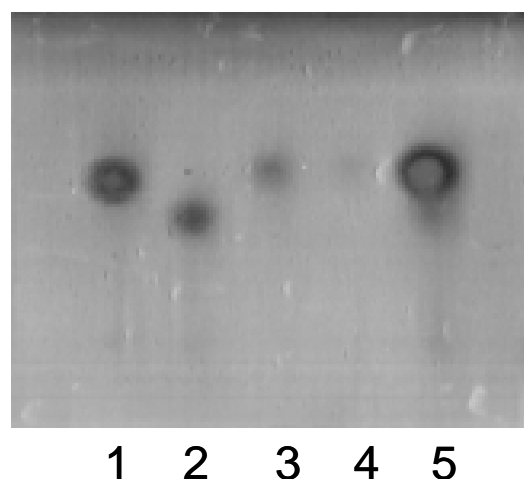


Fig. 1. *Escherichia coli* 215 bioautogram analysis of B₁₂ compounds found in commercially available biofertilizers containing photosynthetic purple bacteria. 1) Authentic B₁₂, 2) authentic pseudo-B₁₂, 3) biofertilizer A concentrated extract (5 μ L), 4) biofertilizer B concentrated extract (5 μ L), and 5) biofertilizer C concentrated extract (2 μ L). Data represent typical bioautograms of three independent experiments.

predominantly formed due to the nucleotide moiety of the molecule.

In the case of authentic Factor III (C₆₁H₈₄CoN₁₄P; monoisotopic mass 1342.5310), this corrinoid was eluted as a peak with a retention time of 7.14 min. MS of authentic Factor III had a major divalent ion at m/z 672.2715 [M+2H]²⁺ (Fig. 2A and B). MS/MS spectrum of Factor III indicated that the ion of m/z 347.0636 was predominantly formed due to the nucleotide moiety (Fig. 2C).

The purified sample was eluted as several total ion peaks, indicating that impurities still existed (Fig. 3A). The ion peaks of m/z 672.27 and 678.29 due to Factor III and B₁₂, respectively, were also found and their retention times were identical to those of authentic Factor III and B₁₂. MS at retention times of 7.14 and 7.35 min showed the formation of both Factor III and B₁₂ divalent ions of m/z 672.2715 (Fig. 3B) and 678.2996 (Fig. 3C), respectively. MS/MS spectrum of each ion peak was identical to that of authentic Factor III and B₁₂ (Fig. 3D and E). These results indicated that biofertilizer C contained both B₁₂ (main) and Factor III (minor).

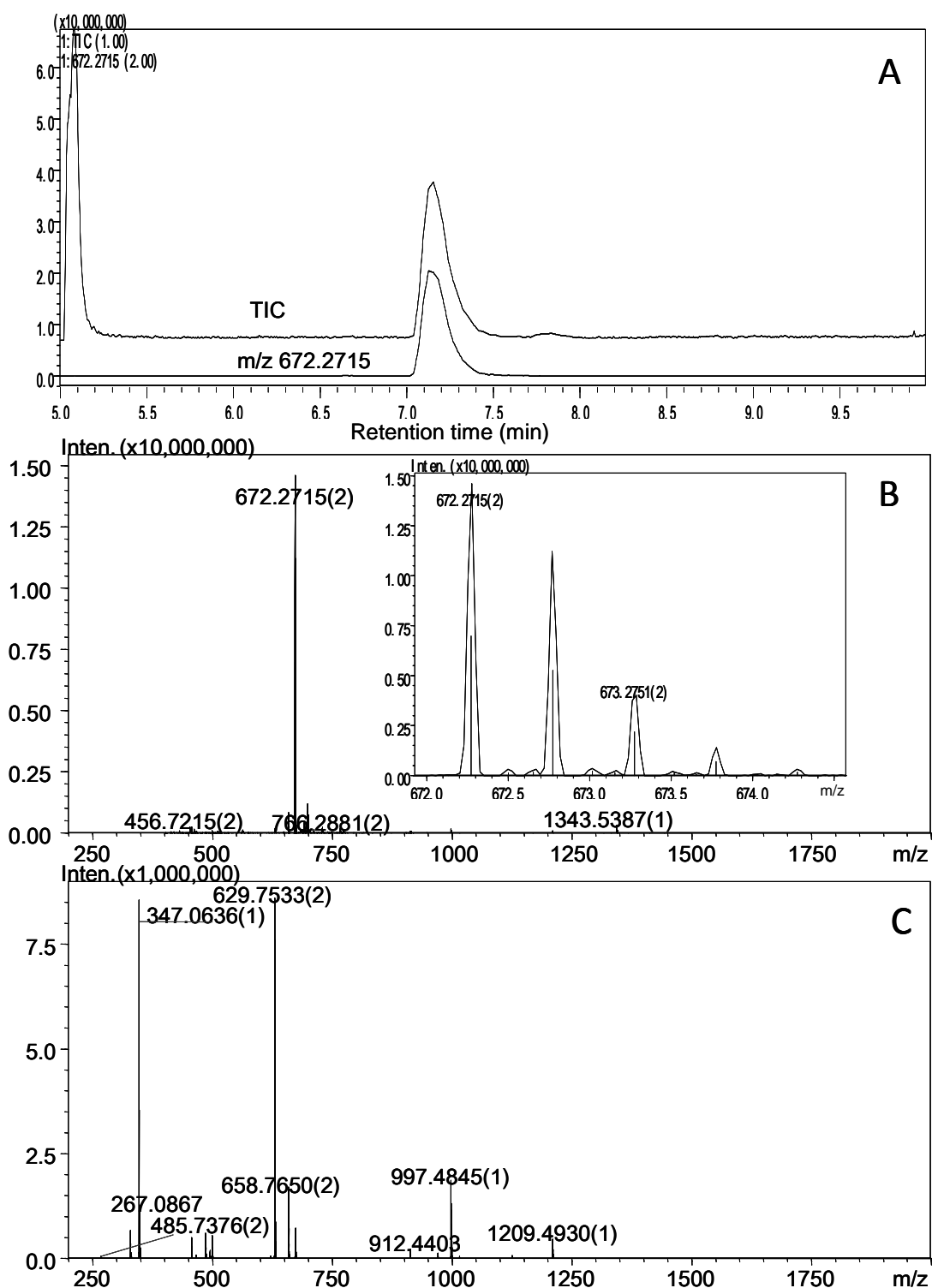


Fig. 2. Liquid chromatography/electrospray ionization mass spectrometry (LC/ESI-MS/MS) chromatograms of authentic Factor III. Factor III was analyzed with LCMS-IT-TOF (Shimadzu) as described in the text. The total ion chromatogram (TIC) of authentic Factor III is shown in panel A. The mass spectrum of authentic Factor III at 7.14 min is shown in panel B; the magnified spectrum from m/z 672 to 675 is inserted in the panel. The MS/MS spectrum for the peak of m/z 672.2715 from authentic Factor III is shown in panel C.

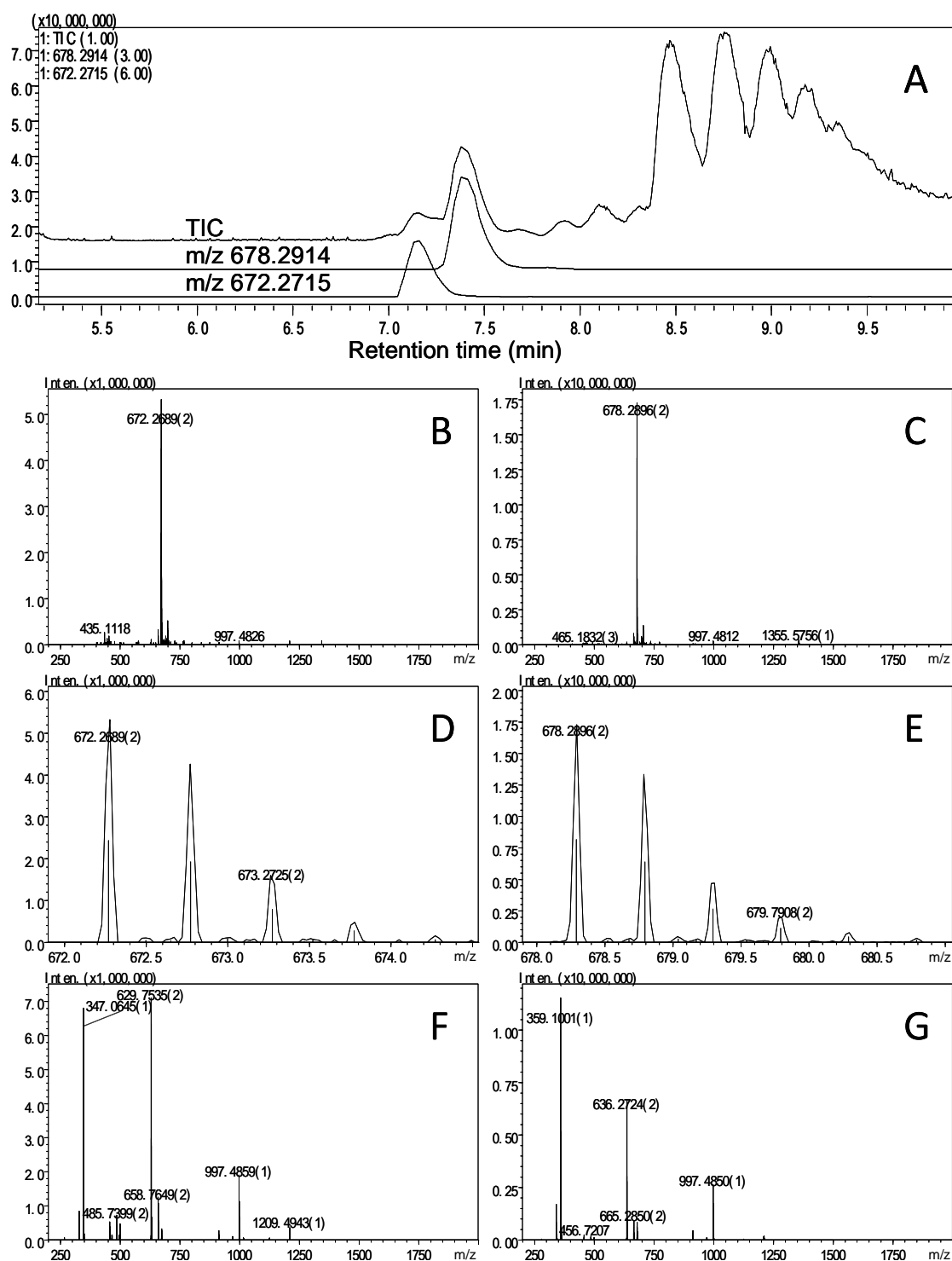


Fig. 3. LC/ESI-MS/MS chromatograms of purified B₁₂ compounds from biofertilizer C. TIC and reconstructed chromatograms of m/z 678.29 and 672.27 of the purified B₁₂ compound are shown in panel A. The mass spectra of the purified B₁₂ compound at 7.14 and 7.30 min are shown in panels B and C, respectively; the magnified spectra from m/z 672 to 675 and m/z 678 to 681 are shown in panels D and E, respectively. The MS/MS spectra for the peak of m/z 672.27 and 678.29 from the purified B₁₂ compound are shown in panels F and G, respectively.

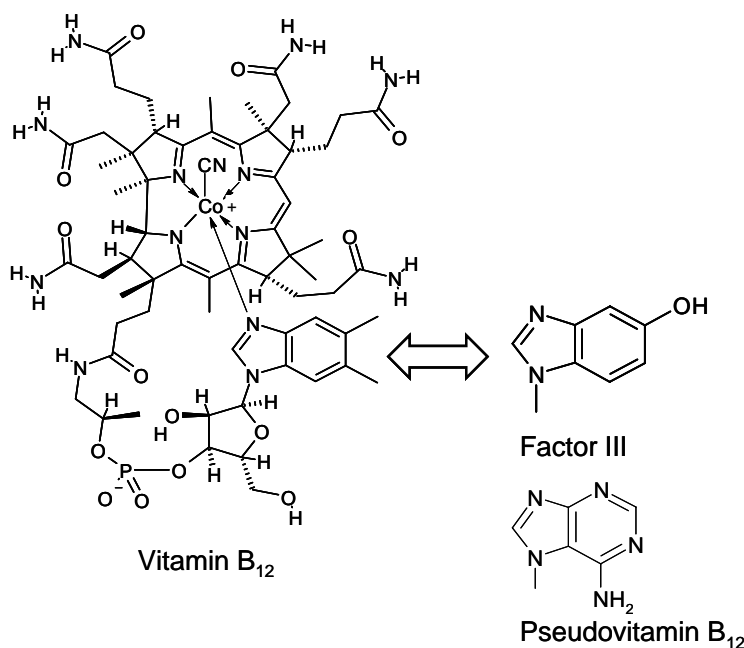


Fig. 4. Corrinoid compounds (vitamin B₁₂ and Factor III) found in biofertilizer C containing purple photosynthetic bacteria. Partial structures of Factor III and pseudo-B₁₂ showing only that portion of the molecule that differs from B₁₂.

Since the N₂-fixing photosynthetic purple bacterium *Rhodobacter capsulatus*, used as a biofertilizer, has the ability to synthesize B₁₂ *de novo* [11], Factor III may be derived from other concomitant bacteria in commercial biofertilizer C (Fig. 4).

Despite the detection of B₁₂ in one of the biofertilizers tested, in our preliminary experiments, B₁₂ was not detected in the lettuce leaves grown with and without biofertilizer C treatment of the soil and leaves once a week for three weeks according to the manufacturer's recommended protocol. These results indicated that commercially available biofertilizers containing purple photosynthetic bacteria are not suitable for B₁₂-enrichment of plants due to the low B₁₂ content.

REFERENCES

- Scheider, Z. and Stroiński, A. 1987, Comprehensive B₁₂, Walter de Gruyter, Berlin, 93.
- Ball G. M. F. 1998, Bioavailability and Analysis of Vitamins in Foods, Chapman & Hall, London, 497.
- Millet, P., Guillaud, J. C., Fuchs, F. and Klepping, J. 1989, Am. J. Clin. Nutr., 50, 718.
- Baik, H. W. and Russell, R. M. 1999, Ann. Rev. Nutr., 19, 357.
- Mozafar, A. 1994, Plant and Soil, 167, 305.
- Allen, R. H. and Stabler, S. P. 2008, Am. J. Clin. Nutr., 87, 1324.
- Mohammadi, K. and Sohrabi, Y. 2012, J. Agric. Biol. Sci., 7, 307.
- Choudhury, A. T. and Kennedy, I. R. 2004, Biol. Fertil. Soils, 39, 219.
- Watanabe, F. 2007, Exp. Biol. Med., 232, 1266.
- Gamal-Eldin, H. and Elbanna, K. 2011, Curr. Microbiol., 62, 391.
- Zappa, S., Li, K. and Bauer, C. 2010, Adv. Exp. Med. Biol., 675, 229.
- Resources Council, Science and Technology Agency. 1995, Standard Tables of Food Composition in Japan-Vitamin K, B₆ and B₁₂. Resources Council, Science, and Technology Agency, Tokyo, Japan, 6.
- Tanioka, Y., Yabuta, Y., Miyamoto, E., Inui, H. and Watanabe, F. 2008, J. Liq. Chrom. Rel. Technol., 31, 1977.
- Hashimoto, E., Yabuta, Y., Takenaka, S., Yamagauchi, Y., Takenaka, H. and Watanabe, F. 2012, J. Nutr. Sci. Vitaminol., 58, 50.